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The branch of chemistry that deals with all non-organic compounds inorganic compounds Forms of an element in the same state with different structures and properties

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Terms in this set (4)
The products of a
reaction between an

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acid and a base are: a)
salt+water b) carbon
dioxide+water. c)
hydrogen+water d)
they not not react.
salt+water. If a
reactant or product in
a chemical equation is
followed by (aq) it
means that the: a)
substance is in
adequate supply.

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Test Multiple Choice

Identify the choice that best completes the statement or answers the question. 1. A

chemical formula includes the symbols of the elements in the compound and subscripts that indicate

- atomic mass of each element.
- number of atoms or ions of each element that are combined in the compound.
- formula mass.

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and Answers 5 The
Periodic Law TEST B 1.
a 2. c 3. d 4. d 5. a 6. a
7. c 8. a 9. lanthanides
10. 2 11. fourth 12.
transition elements 13.
32 14. valence
electrons 15. electron
affinity 16.

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electronegativity 17.
ionization energy 18. 3
s² 3p⁴ 19. atomic
radius 20 ...

Compounds

Chapter 7 Chemistry Test Answer Key

A balanced equation represents the law of mass conservation, and as a result all of the following are true except ____.

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Chapter 7. energy and
Chemical reactions.

249. energy...it makes
things happen. To get
an idea of the role
energy plays in our
lives, let's spend some
time with John, a
college student in one

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of the coastal towns in California. He wakes up in the morning to a beautiful sunny day and decides to take his chemistry book to the beach.

Chapter 7 - An Introduction to Chemistry: Energy and ...

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Test Answer: CH₃ - CH₂Cl, CH₃ - CHCl₂, CH₃ - CCl₃, CH₂Cl - CCl₃, CHCl₃ - CCl₃, CCl₃

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- CCI 3. Chapter 7

Chemical Reactions

Question 3. Write down the structural formulae of ethane and ethene.

Answer: $\text{CH}_3 - \text{CH}_3$ -

Ethane $\text{CH}_2 = \text{CH}_2$ -

Ethene. Methyl

Propanoate Question 4.

What is the peculiarity of the Carbon- Carbon bond in ethene?

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help you understand the fundamentals of chemical reactions (topic 7 of IGCSE chemistry syllabus).

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IGCSE CHEMISTRY REVISION [Syllabus 7]- Chemical Reactions

When a reaction system is at equilibrium a) there is

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no more chemistry happening. b) the amounts of reactants and products are exactly equal. c) the reaction rate in the forward direction is at a maximum. d) the reaction rate in the reverse direction is at a minimum. e) the rates...

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explanation of some
chemical reaction,

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differentiate between
physical changes and
chemical reactions,
classify the chemical
reactions-...

Chemical Reaction | SSC Chemistry Chapter 7 |

□□□□□□□□ □□□□□□□□ |
Fahad Sir

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and formulas in the MA
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Who doesn't love
pouring two different
chemicals in one test
tube and see the
Changes of the
properties? ... Chemical
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Flashcard. A chemical
formula includes the
symbols of the
elements in the
compound and
subscripts that
indicate. Number of
atoms or ions of each
element that are
combined in the
compound. A chemical
formula for a molecular
compound represents
the composition of. A

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molecule.

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7.2: Chemical
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Equations A chemical reaction is the process in which one or more substances are changed into one or more new substances. Chemical reactions are represented by chemical equations. Chemical equations have reactants on the left, an arrow that is read as "yields", and the products on the right. 7.3: How to Write Balanced Chemical ...

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